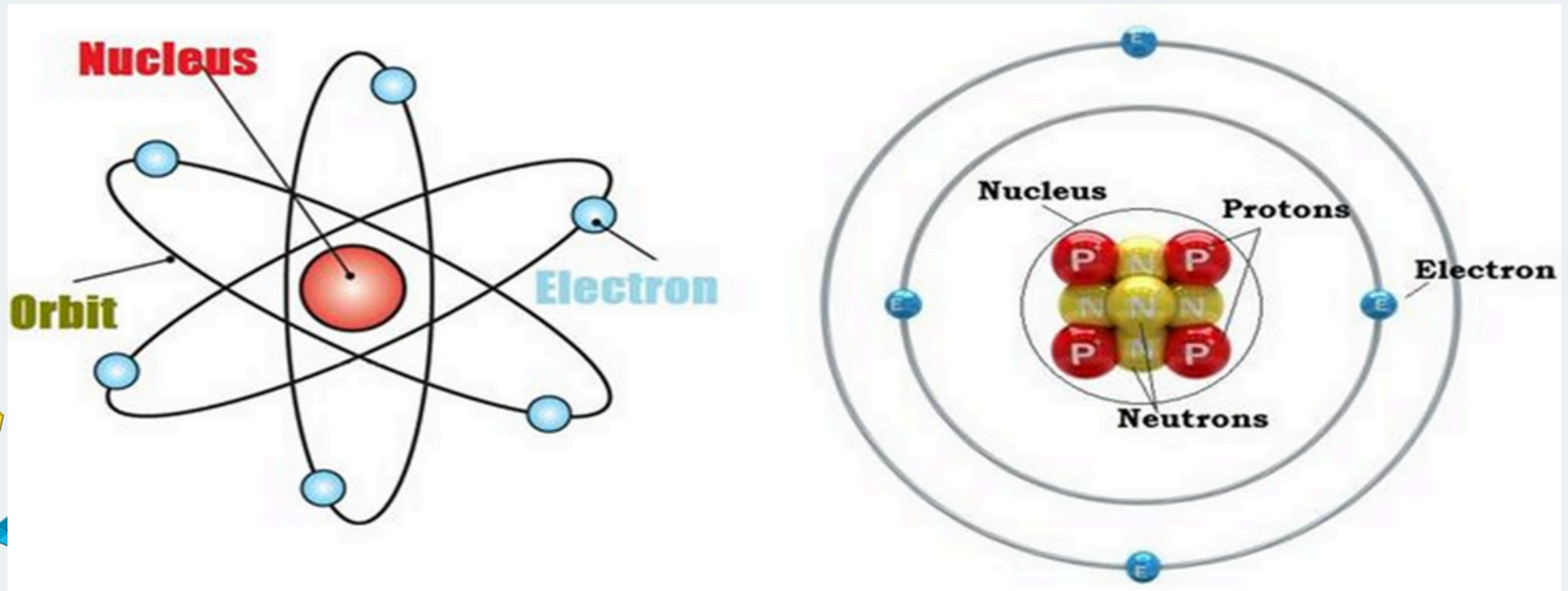


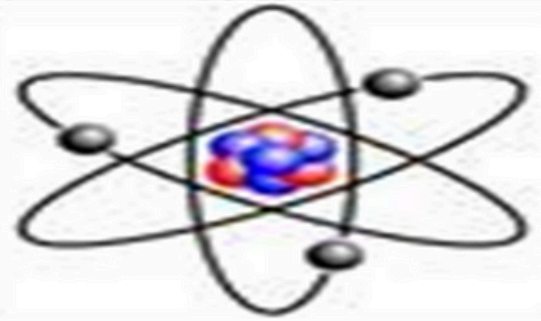
Introduction to Electric in Petrochemical

ATOM STRUCTURE



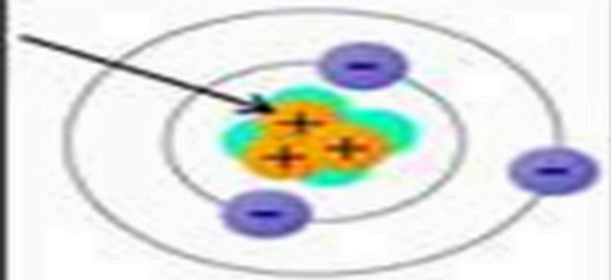
Atom

The smallest unit of matter that can't be broken down chemically



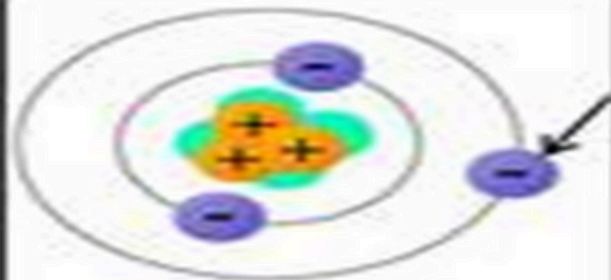
Proton

A positively charged subatomic particle



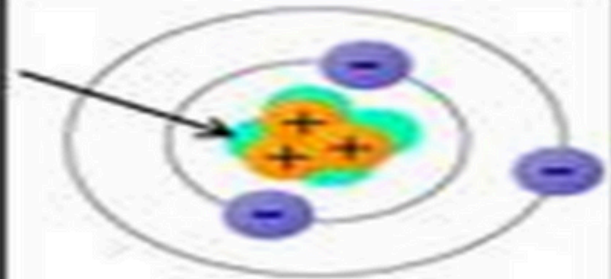
Electron

A negatively charged subatomic particle



Neutron

A subatomic particle with no charge



Conductor – high electrical conductivity, is
metallic element usually in group 1 until 3

The Metals

The Metals

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Semiconductor – is intermediate between metal & non-metal (group 4)

Table of the Elements

2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg	IIIB	IVB	VB	VIB	VIB	— VII —					13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4	19 K	20 Ca	21 Sc	22 Ti	23 Y	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra	+Ac	104 Rf	105 Ha	106 Sg	107 Ns	108 Hs	109 Mt	110	111	112	113					

Semiconductor materials

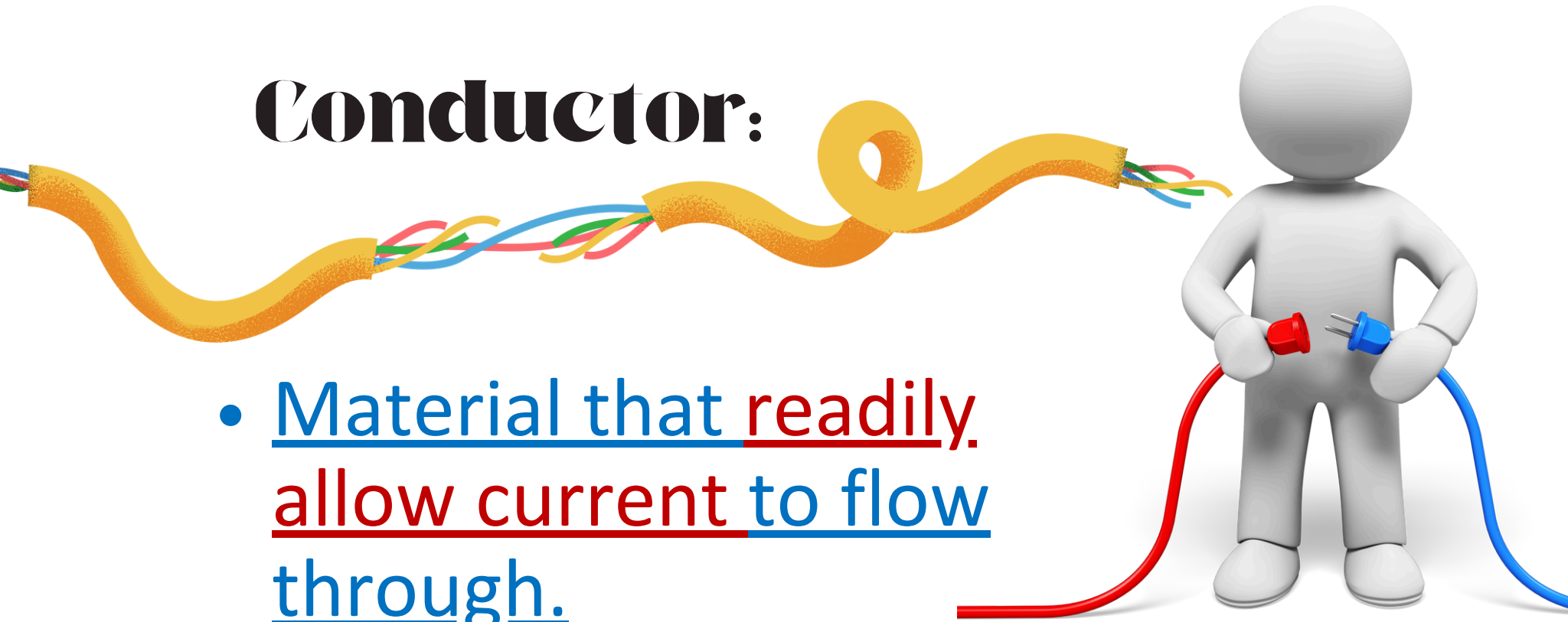
* Lanthanide Series

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

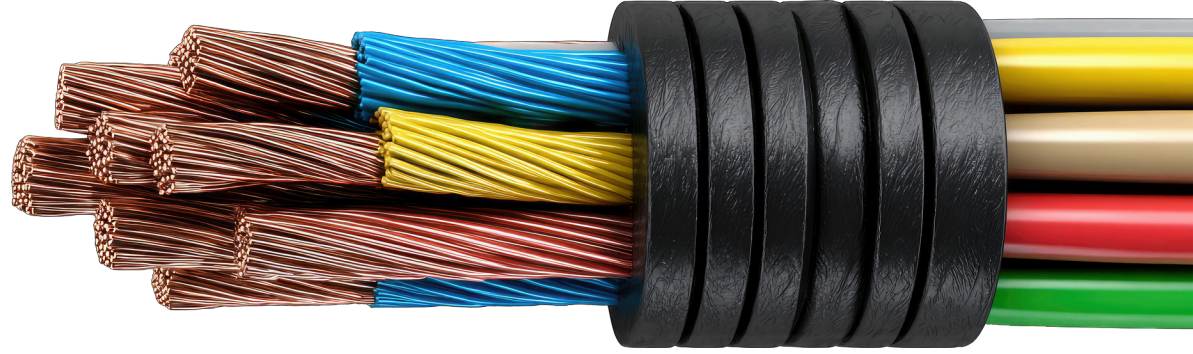
+ Actinide Series

Conductor:

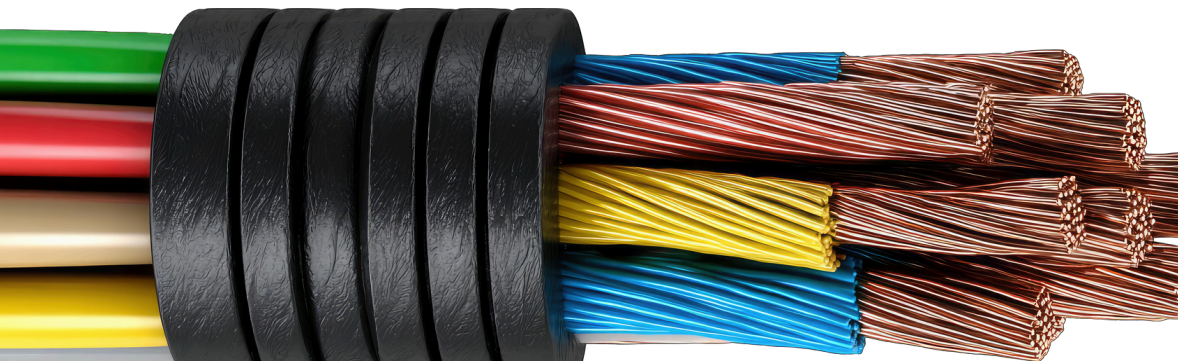
- Material that readily allow current to flow through.
- Generally have high electrical conductivity when in solid state



Insulator:



- Material that poor conductors electric current, does not conduct electrical current under normal condition.
- Used to prevent current when it is not wanted.



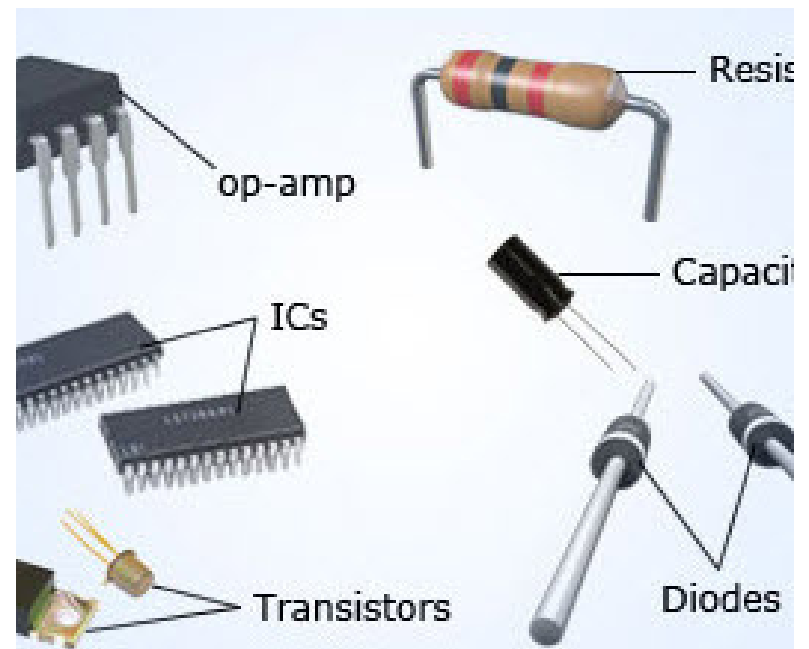
Semi Conductor:

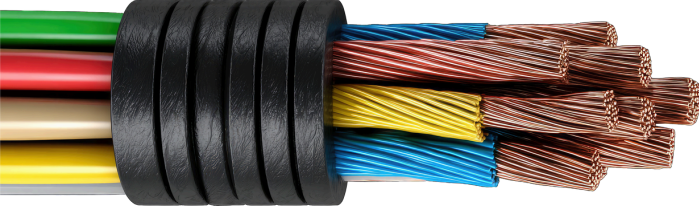
- Material that is between conductor & insulator in its ability to conduct electric current.
- It classed below the conductor in their ability to carry current



SEMI CONDUCTOR:

Unique
characteristic ,
certain
semiconductors
material are basis
for electronic
device



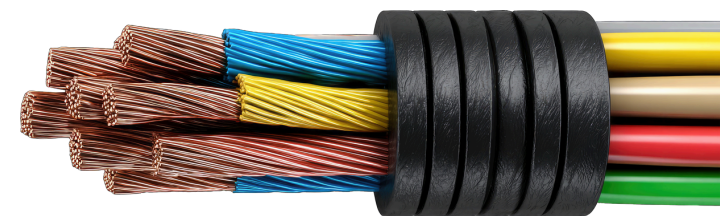


Conductor (1-3)

Semiconductor

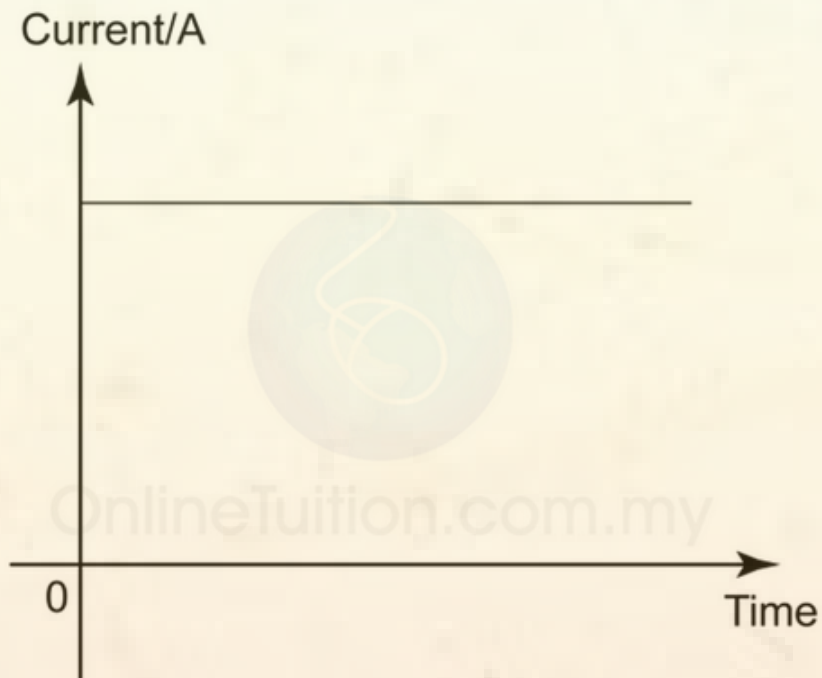
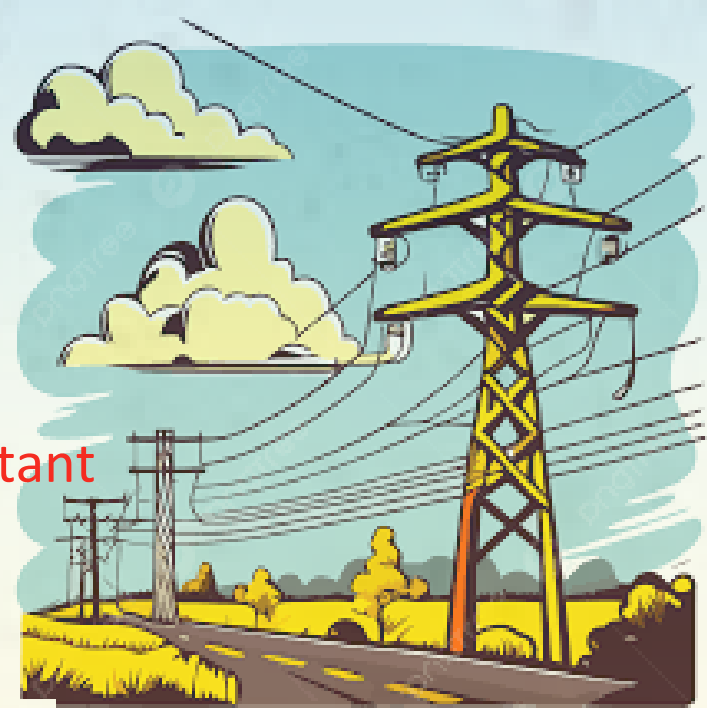
Insulator (5-8)

Periodic Table of the Elements																																					
1 IA 1A		2 IIA 2A										13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A	17 VIIA 7A	18 VIIIA 8A																				
1 H Hydrogen 1.008		3 Li Lithium 6.941	4 Be Beryllium 9.012									5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180																				
11 Na Sodium 22.990	12 Mg Magnesium 24.305			3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIB 7B	8 VIII 8		9 VIII 9	10 VIII 10	11 IB 1B	12 IIB 2B	13 IIIB 3B	14 IVB 4B	15 V 5A	16 VI 6A	17 VII 7A	18 VIII 8A																	
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80	37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.905	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29		
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71		72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.384	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium [209]	85 At Astatine [210]	86 Rn Radon [222]	87 Fr Francium [223]	88 Ra Radium [226]	89-103		104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]	107 Bh Bohrium [264]	108 Hs Hassium [277]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [271]	111 Rg Roentgenium [272]	112 Cn Copernicium [285]	113 Uut Ununtrium [284]	114 Fl Flerovium [289]	115 Uup Ununpentium [288]	116 Lv Livermorium [293]	117 Uus Ununseptium [294]	118 Uuo Ununoctium [294]



DIRECT CURRENT

- Flowing continued to **one direction only**
- **the polarity** to the terminal voltage cell is **constant**



BATTERY

- Combination of cell
- Cell - one unit or a combination of material that is used to **convert chemical energy into electrical energy**
- Output current depend on output voltage and resistance

$$I = \frac{V}{R}$$

I = current

V = voltage

R = resistance



ELECTROMOTIVE FORCE, E.M.F

- Electric current cannot flow through the conductor until one source of external energy supply to that conductor
- It provided by a source of energy such as a battery or generator and measured in volts
- e.m.f is the external work expended per unit of charge to produce an electric potential difference across 2 open circuit terminal
- Known as measurement of energy that causes current to flow through circuit - voltage



Charge

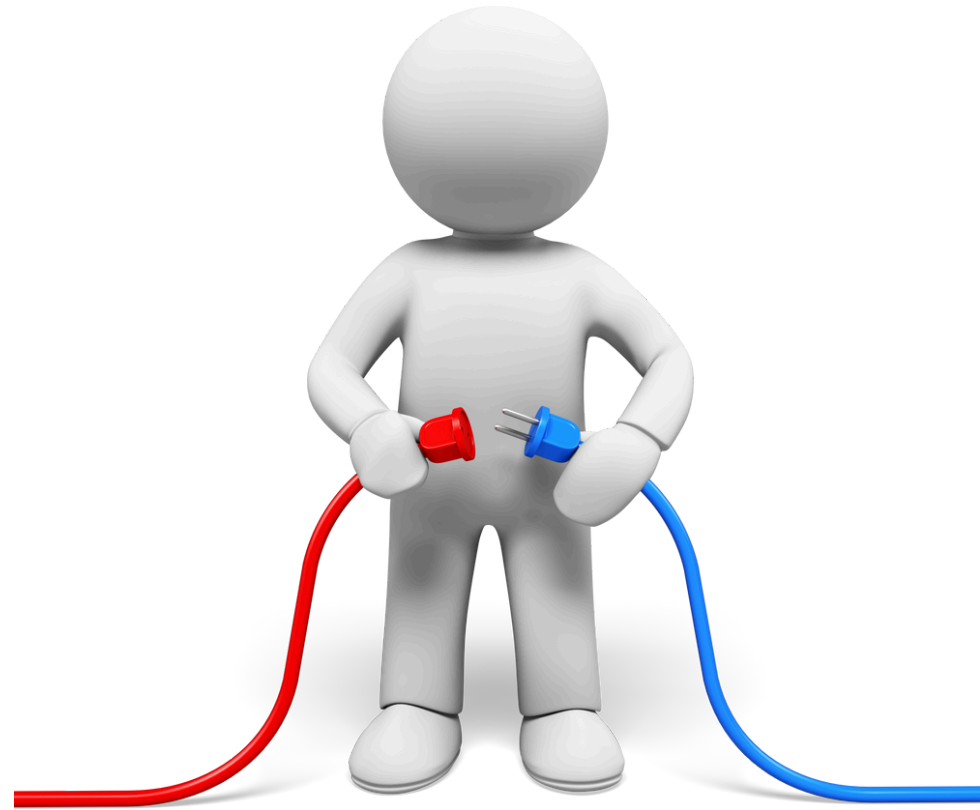
- Unit of charge is the coulomb (C) where one coulomb is one ampere second
- 1 Coulomb = 6.24×10^{18}

$$Q = It$$

Q – charge

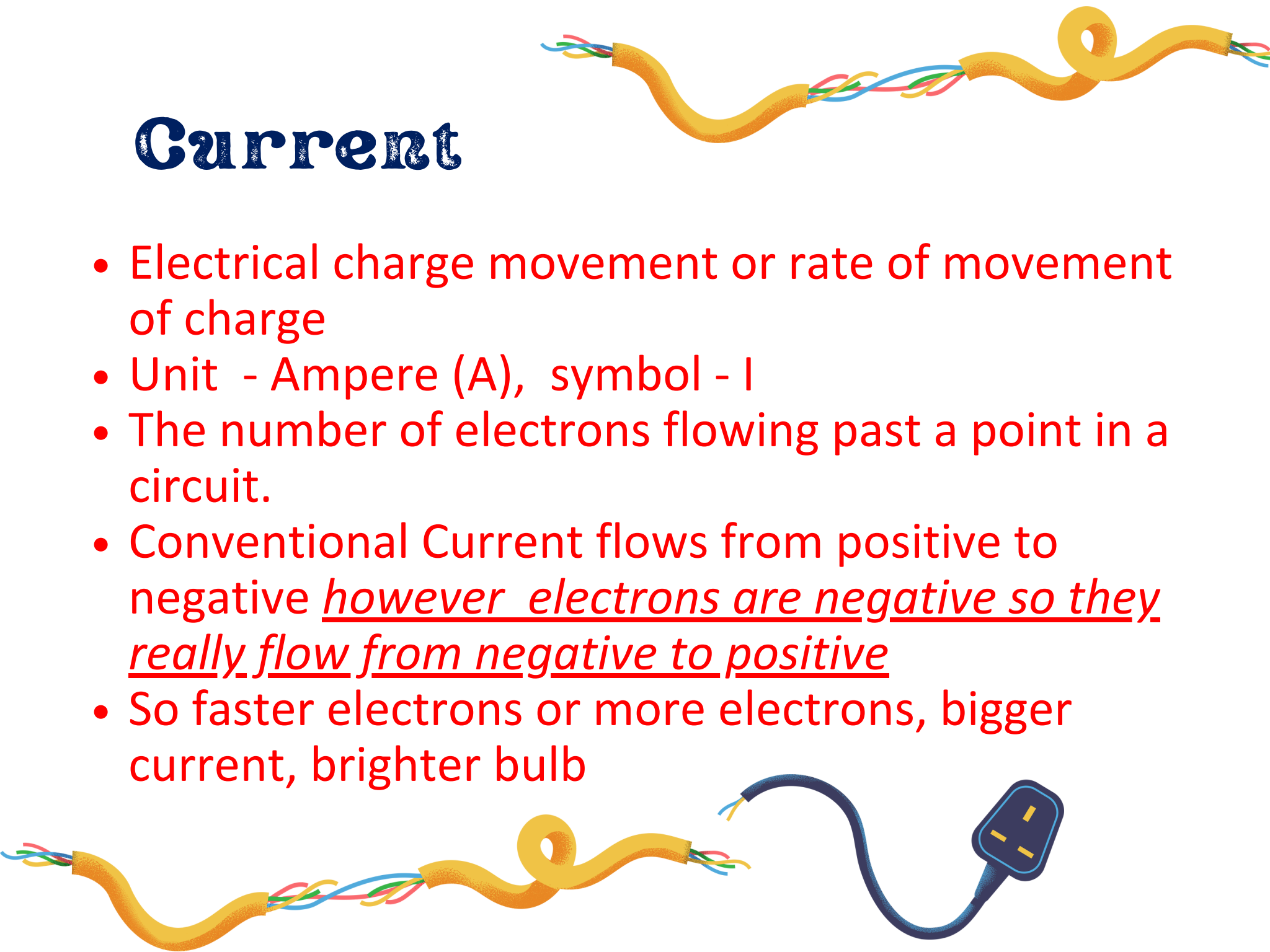
I - current in ampere

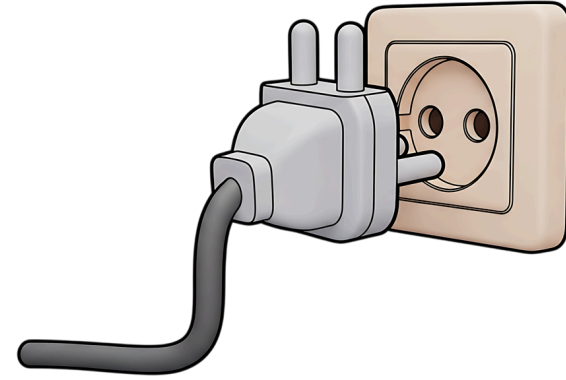
t - time (second)



Current

- Electrical charge movement or rate of movement of charge
- Unit - Ampere (A), symbol - I
- The number of electrons flowing past a point in a circuit.
- Conventional Current flows from positive to negative however electrons are negative so they really flow from negative to positive
- So faster electrons or more electrons, bigger current, brighter bulb

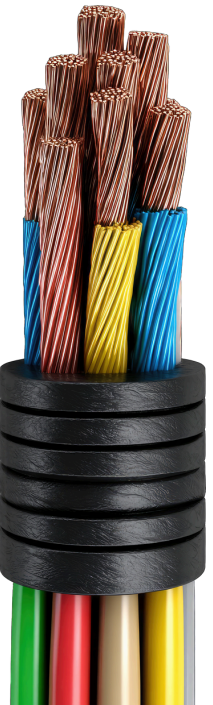




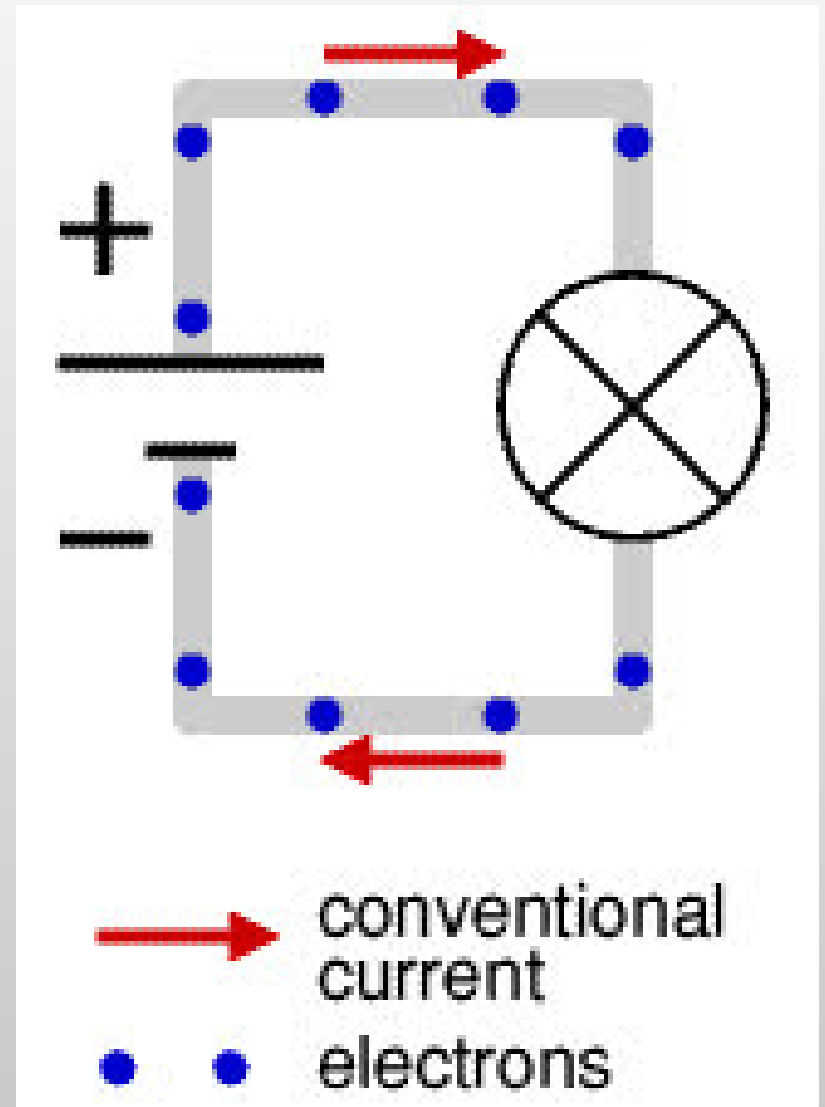
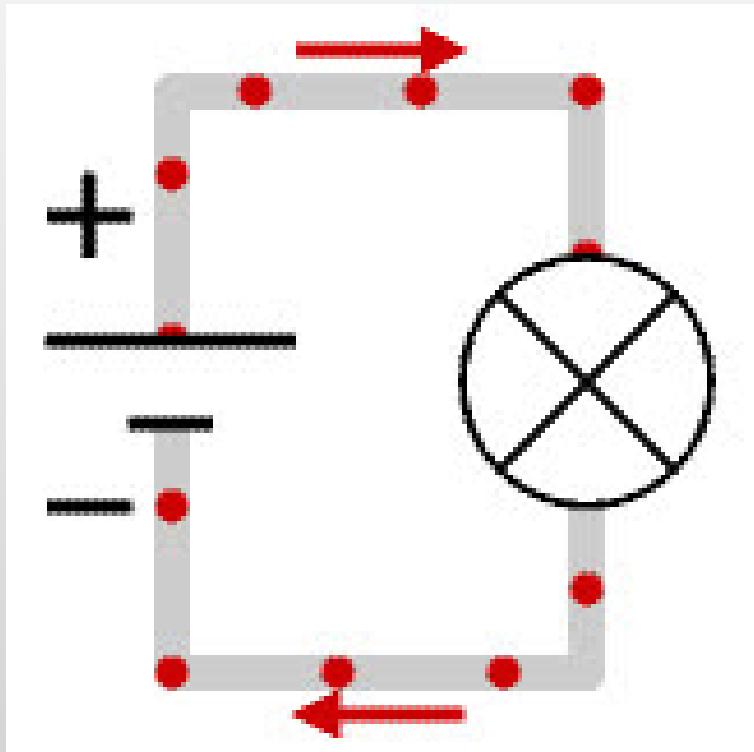
- Symbol: I
- Units: ampere (A)
(ampere = amps)

$$I = \frac{Q}{t}$$

1 ampere is the amount of current that exists when a number of electrons having a total charge of one coulomb move through a given cross-sectional area in one second



REPRESENTING CURRENT DIRECTION

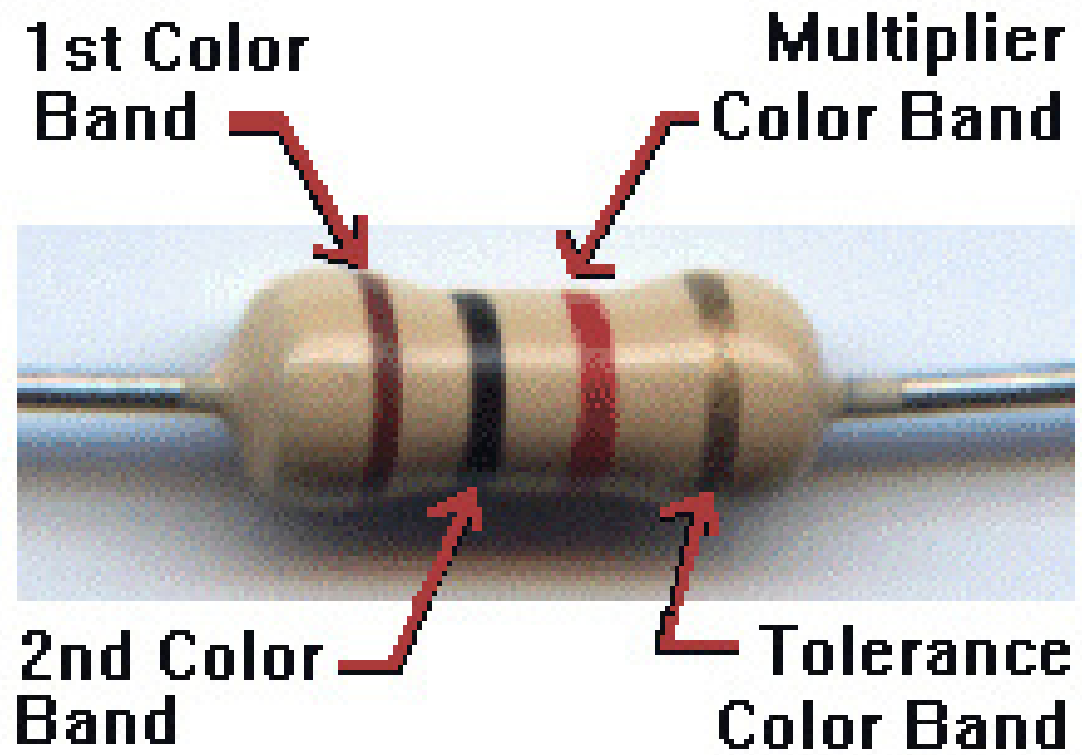


Potential energy (Voltage)

- Define as the difference in energy level of a unit charge located at each of 2 points in a circuit
- Represents the energy required to move the unit charge from one point to the other
- Known as electrical potential, unit volt (V)
$$\text{volts} = \frac{\text{watts}}{\text{ampere}} \quad \text{OR} \quad \text{volts} = \frac{\text{joules}}{\text{coulombs}}$$

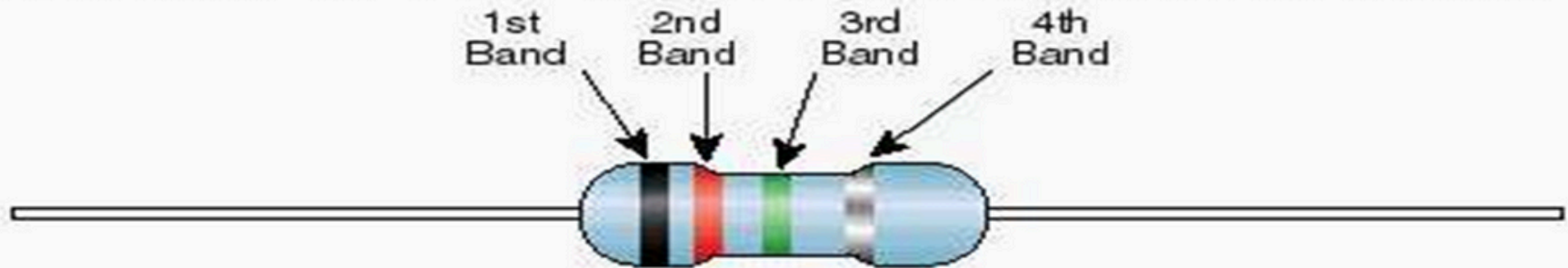
Resistance

- Define as resistance that exist from conductor in current flow
- Unit is ohm (Ω) and symbol is R
- Opposition of current in electric circuit
- One ohm (1Ω) of resistance if there is one ampere ($1A$) of current in a material when one volt ($1V$) is applied across the material

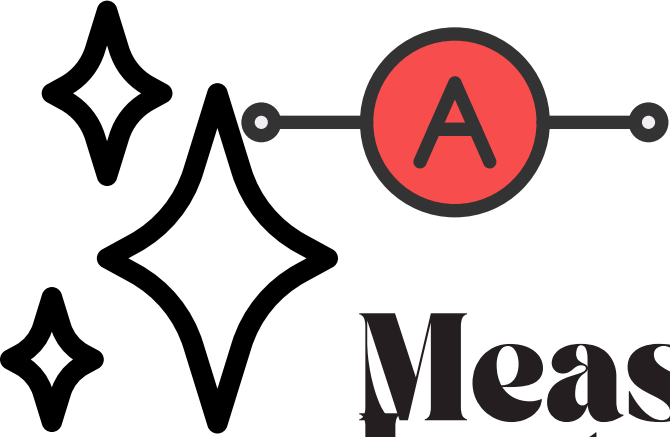


Resistor Colour Code:

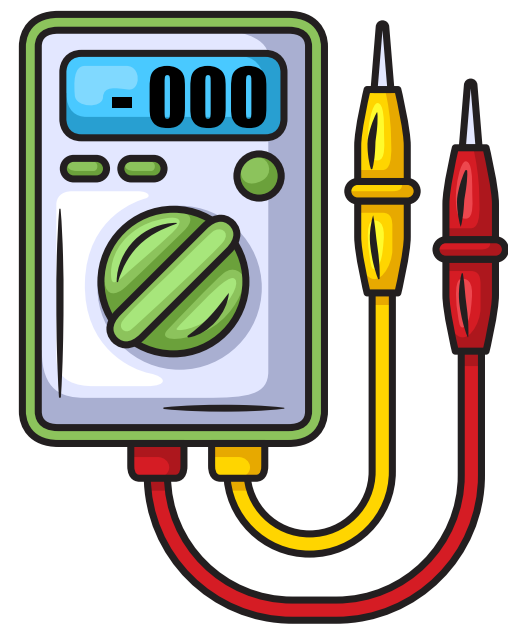
Standard EIA Color Code Table 4 Band: $\pm 2\%$, $\pm 5\%$, and $\pm 10\%$



Color	1st Band (1st figure)	2nd Band (2nd figure)	3rd Band (multiplier)	4th Band (tolerance)
Black	0	0	10^0	
Brown	1	1	10^1	
Red	2	2	10^2	$\pm 2\%$
Orange	3	3	10^3	
Yellow	4	4	10^4	
Green	5	5	10^5	
Blue	6	6	10^6	
Violet	7	7	10^7	
Gray	8	8	10^8	
White	9	9	10^9	
Gold			10^{-1}	$\pm 5\%$
Silver			10^{-2}	$\pm 10\%$

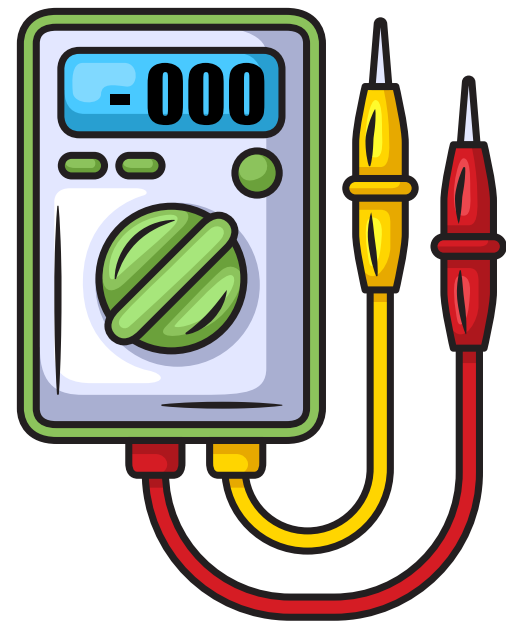


Measuring Instruments



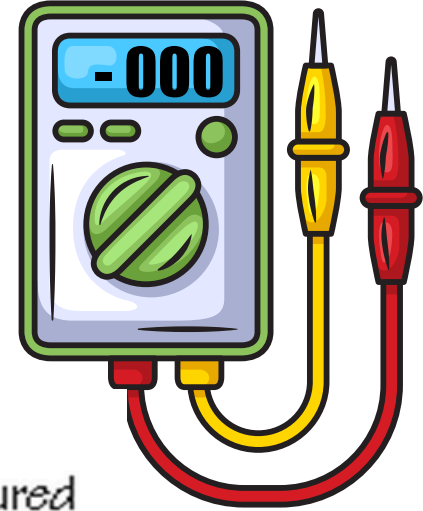
There are many types of electronic measuring instrument. Such as Wattmeter, Ohmmeter, Voltmeter, Ammeter and Megger

The Ohmmeter



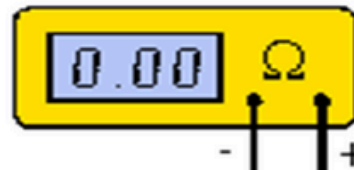
- Symbol Ω
- Instrument for measuring electrical resistance
- Does not function with a circuit connected to a power supply
- To measure the resistance of particular component, it must take out from the circuit and tested separately

How to connect Ohmmeter



ohmmeter
200 kilohms fcd

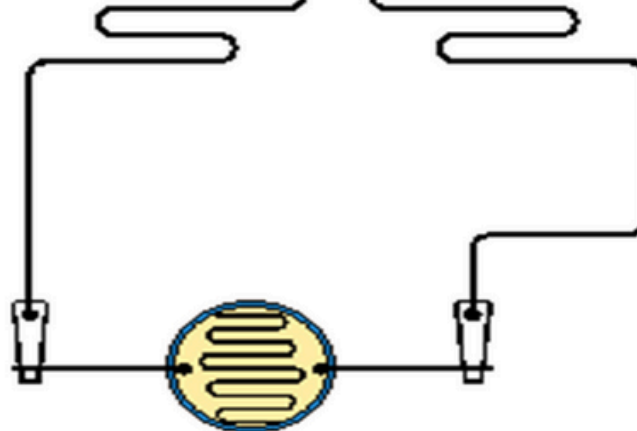
resistance may only be measured
when components are removed
from any circuit



black lead

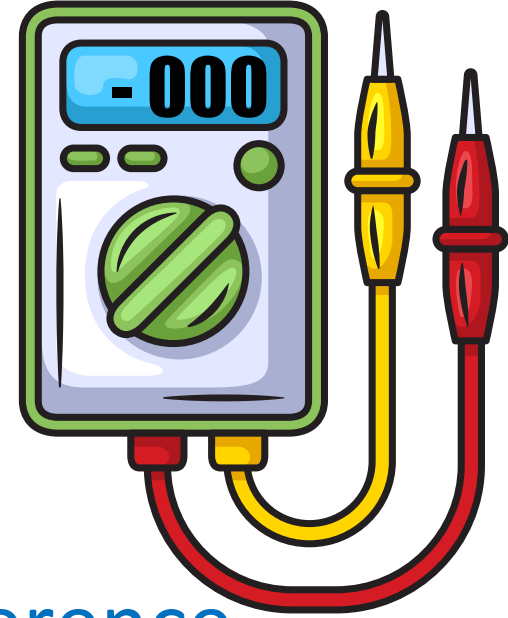
red lead

crocodile
clip



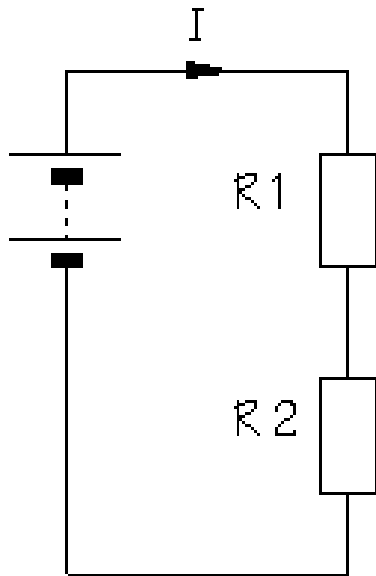
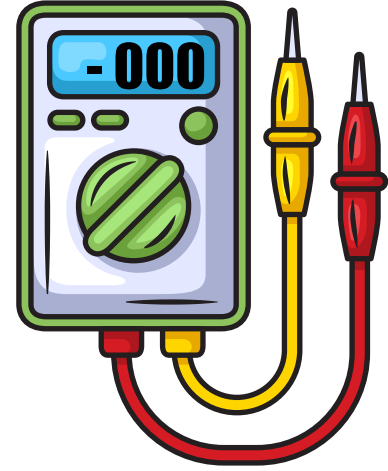
LDR

Voltmeter

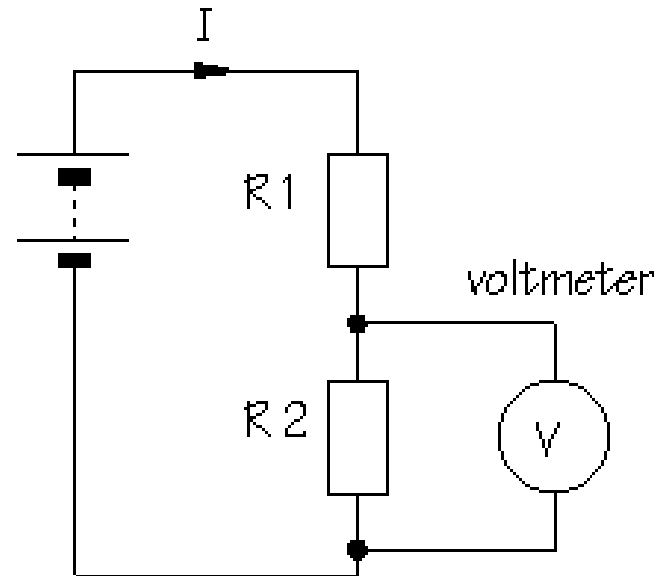


- symbol V
- Used to measure the potential difference (voltage) between 2 points
- The circuit is not changed, hence voltmeter is connected in parallel
- Voltmeter should have a very HIGH resistance

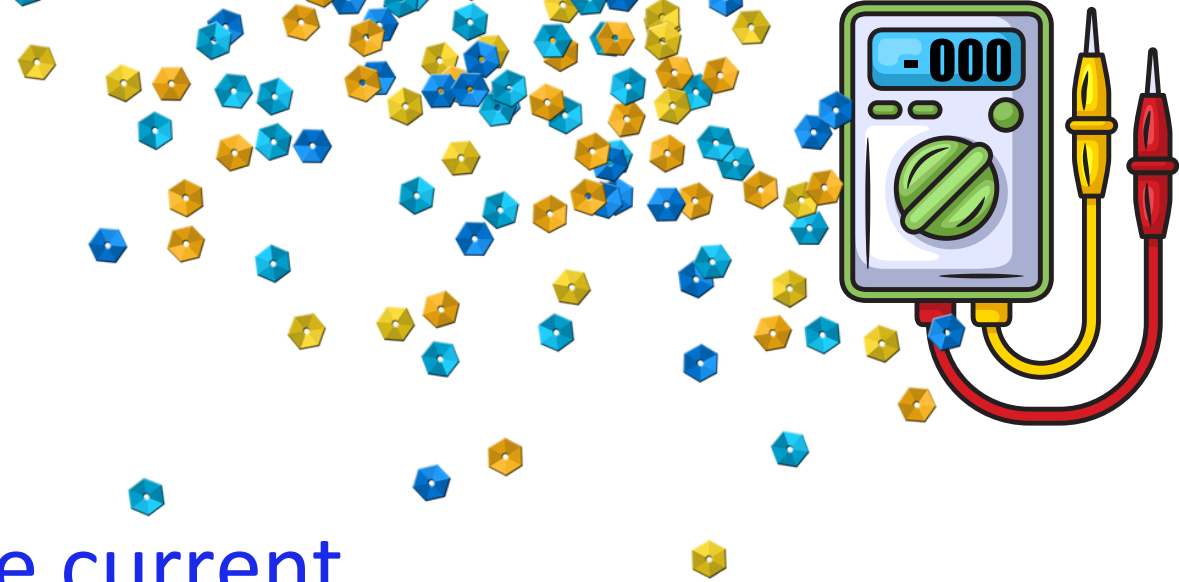
How to connect Voltmeter



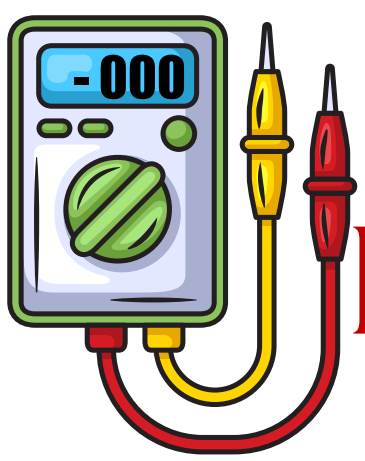
V



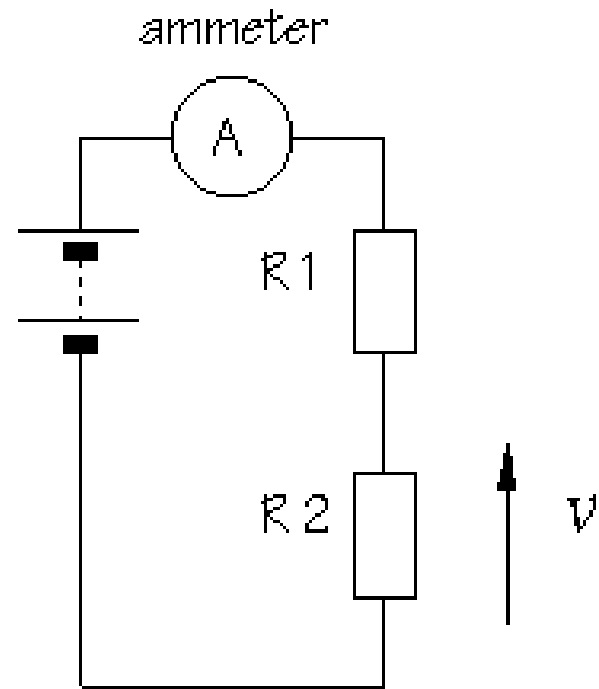
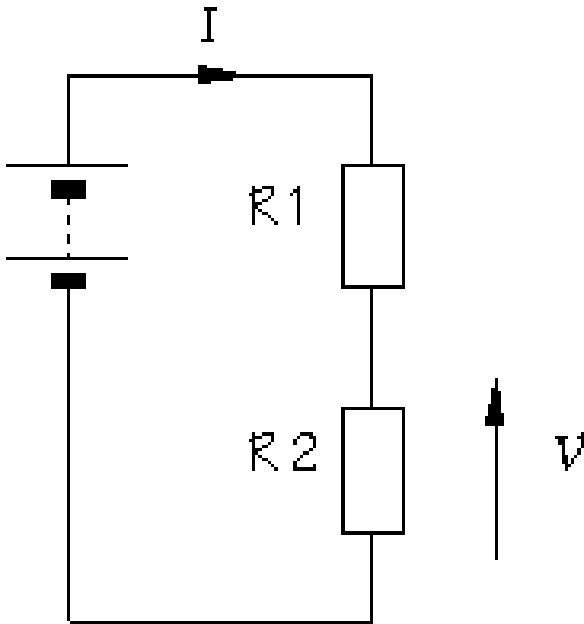
Ammeter



- Symbol A
- Used to measure current
- To measure current, the circuit must be 'broken' ,to allow the ammeter to be connected in series
- Ammeter must have a low resistance
- All the current flowing in the circuit must through the ammeter



How to connect Ammeter





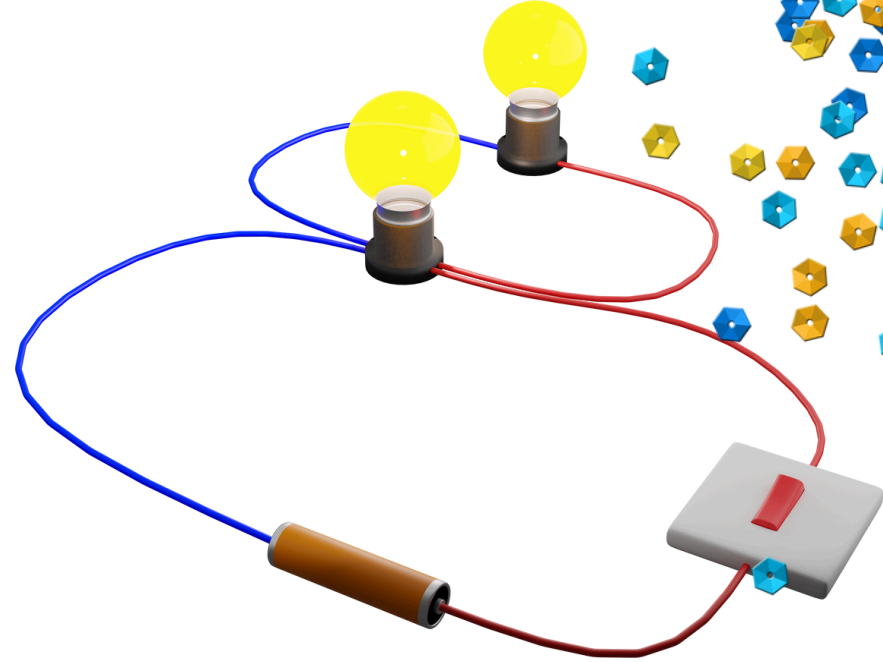
Types of Electrical Circuit

Simple Circuit

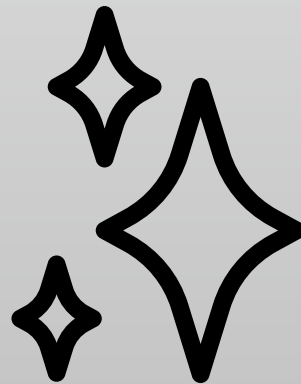
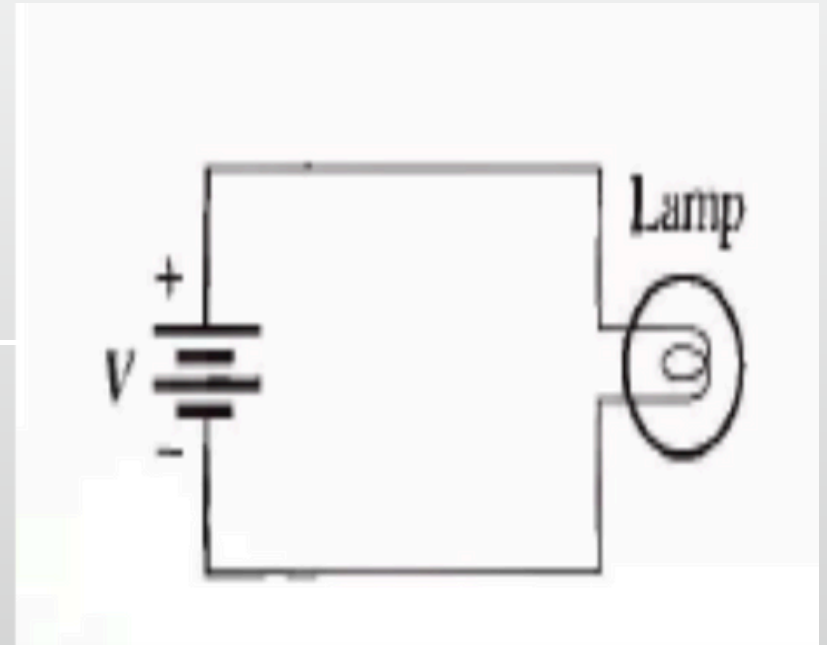
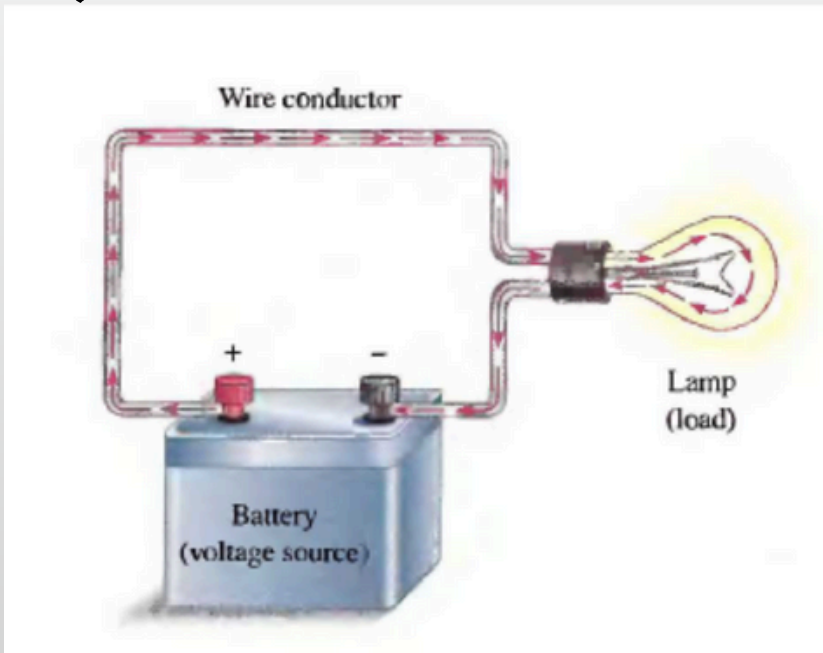
Consist of 3 basic things:

- Voltage source
- Load
- Path for current

A simple circuit consists of a source of potential difference (battery), a load or device that uses the energy (light, stereo, etc), and conductors creating a continuous path (wires).



Simple Circuit



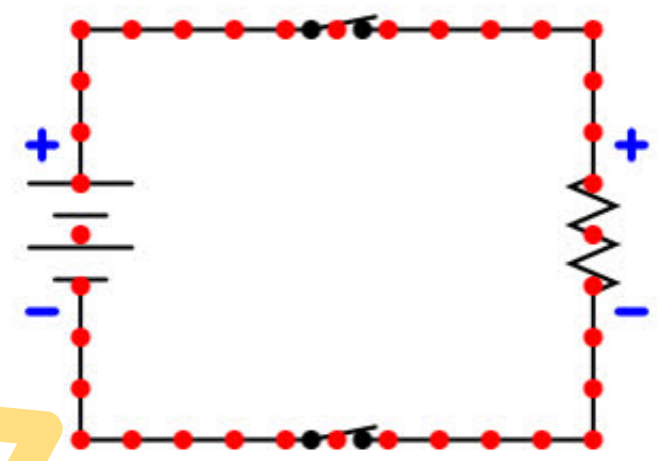
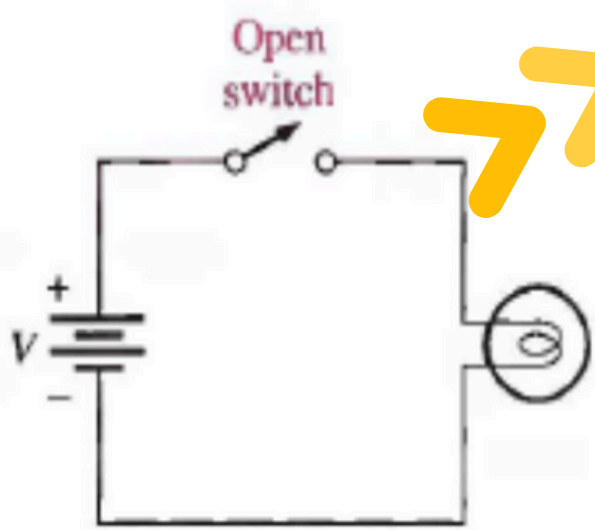
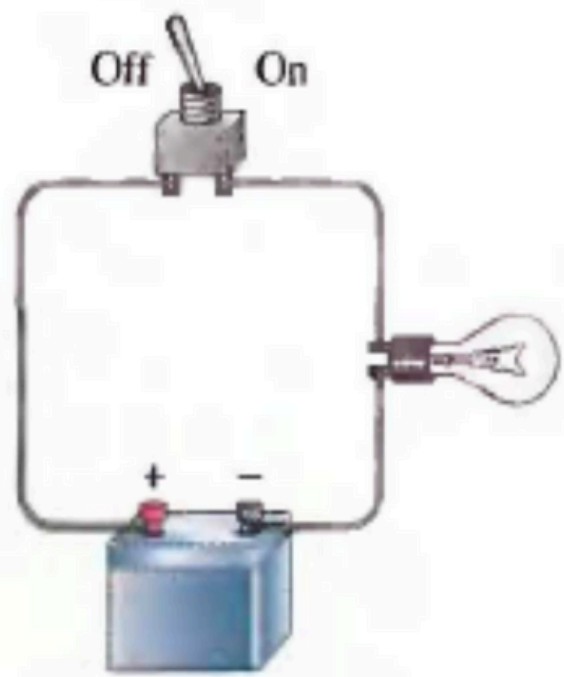
Open Circuit

- The current path is broken
- A condition in an electric circuit in which there is no path for current between 2 points
- Example: broken wire or a switch in the 'open' or 'off' position



Direction of electron motion

Open circuit

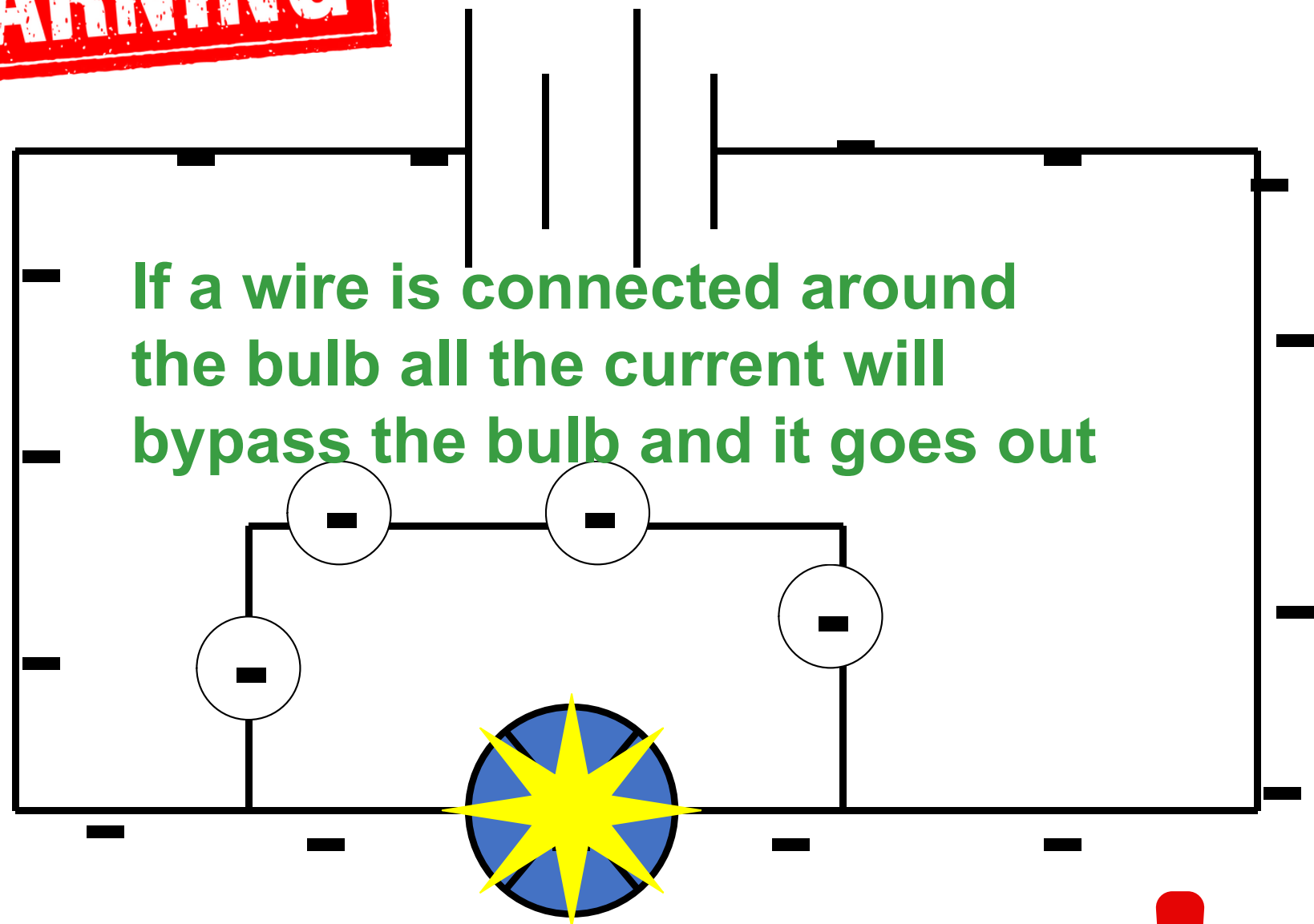


Short Circuit

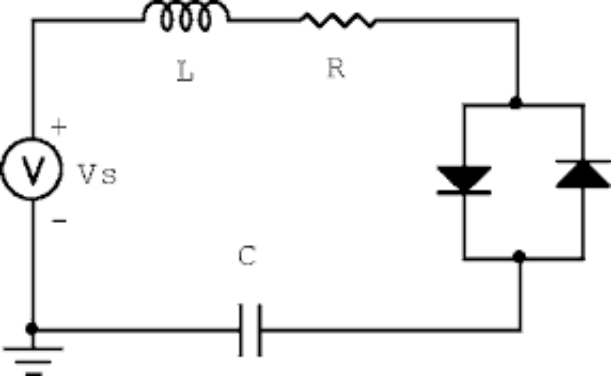


- A low resistance connection established by accident or intention between 2 points in the circuit
- The current tends to flow through the area of low resistance, bypassing the rest of the circuit
- The voltage tends to zero, and current tends to infinity

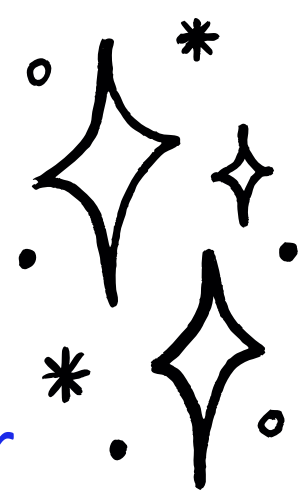
WARNING



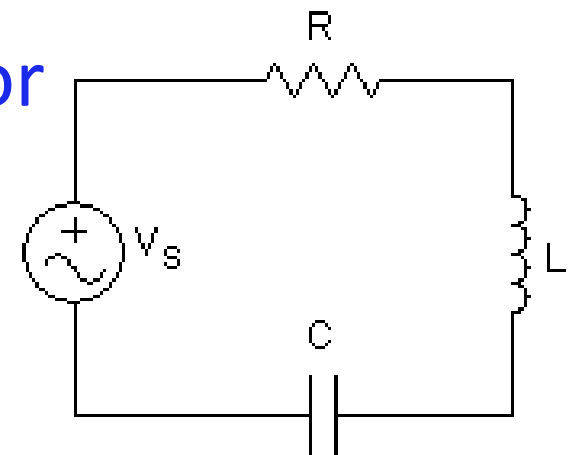
We call this a SHORT CIRCUIT !!

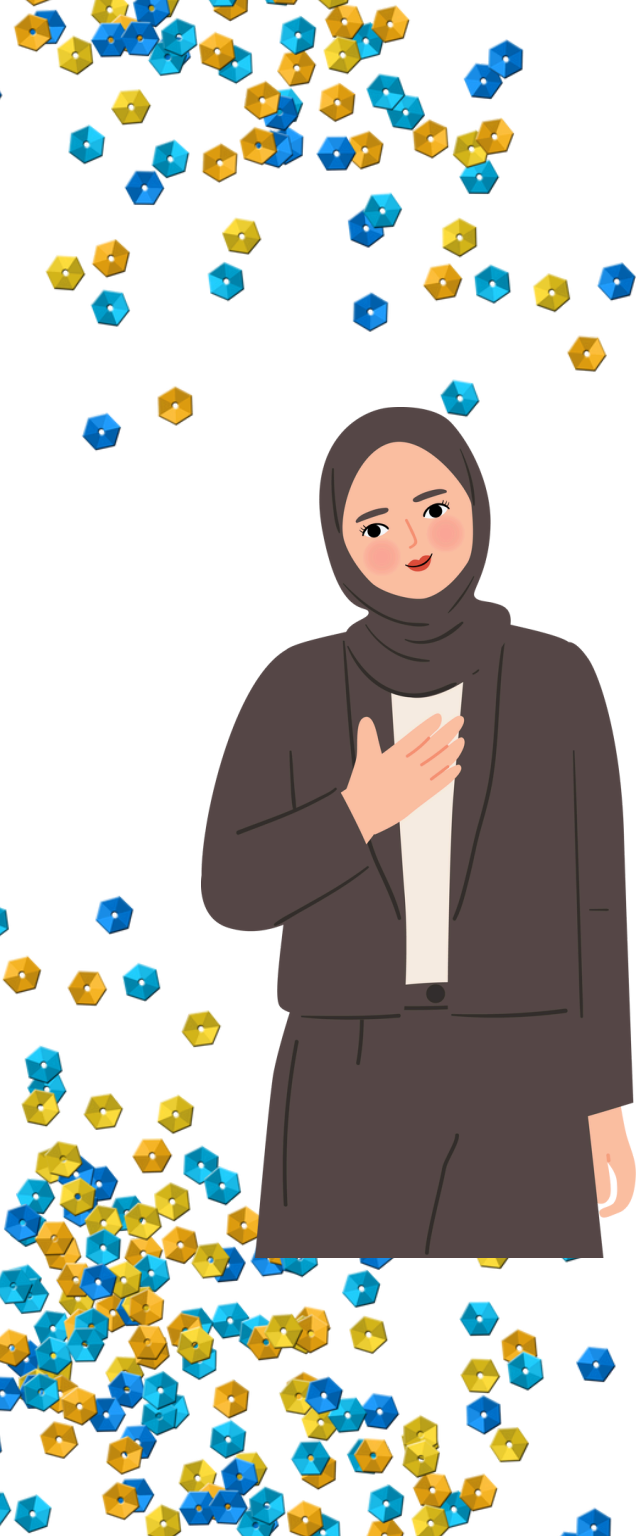


Complex Circuit



- A circuit that consist of inductor (L), resistor (R) and Capacitor (C) or diode
- All the components are treated as resistor to find net impedance
- Complex number must sometimes be used to solve this types of circuit
- contains RLC and diode or transistor in neither purely series or parallel.





Thank
You!